

Biomolecular-chemical Screening
A Novel Screening Approach for the Discovery of
Biologically Active Secondary Metabolites

II. Application Studies with Pure Metabolites[†]

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(Received for publication September 1, 1999)

The novel screening strategy called "biomolecular-chemical screening" combines the advantages of the chemical screening approach—the analysis of the chromatographic and chemical behaviour of secondary metabolites on TLC plates—with binding studies of these molecules with bio-macromolecules like DNA. This approach was advantageously used to detect the interaction of pure compounds with DNA. In order to prove the reliability of the biomolecular-chemical screening and to examine DNA-binding properties, 470 pure secondary metabolites were analysed by this method. Besides the confirmation of already known binders with the TLC-based method, for a number of natural products DNA-binding properties were discovered for the first time. In consequence, binding of pure compounds can be measured by 1D TLC in a reliable and easy manner, in which DNA is applied together with the test compound at the starting spot. Analysis is performed *via* differences in R_f-values in comparison to a reference chromatogram without DNA.

It is generally accepted that nature provides a broad spectrum of structurally diverse metabolites which are expected to serve as new lead structures for pharmacological and agricultural applications¹⁾. For anti-cancer and anti-infective treatment, between 1989 and 1995 even 60% of the approved drugs and the pre-new drug application candidates (excluding biologics) have been of natural origin²⁾. Beside topoisomerase inhibitors and tubulin interacting agents, DNA-binding compounds are in current use and as such play an important role in cancer therapy. Actinomycin D, bleomycin A₂, calicheamicin γ_1 , doxorubicin, mitomycin C and rebeccamycin are low molecular weight microbial metabolites that interact with the DNA double helix, damage DNA or prevent correct replication. In addition, antisense strategies covering the field of therapeutic adjustment of gene expression or

selective inhibition of a virus or microorganism also make use of DNA-binding compounds to stabilize nucleic acid triplex complexes *in vivo*³⁾. Because of the ongoing interest, considerable research is underway to expand the spectrum of compounds with different modes of DNA-binding.

In the previous paper we described a novel screening approach called biomolecular-chemical screening in which the interaction of low molecular weight substances with DNA can be determined by means of thin-layer chromatography⁴⁾. In this paper we report on DNA-binding properties of 470 pure natural compounds from our laboratory stock which were examined with the new one-dimensional TLC method. This investigation aimed at both, to prove the reliability of the method developed, and to discover natural products with hitherto unknown DNA affinity for additional biological profiling.

[†] Dedicated to Prof. Dr. AXEL ZEECK on the occasion of his 60th birthday.

Materials and Methods

General

To obtain random DNA, salmon sperm DNA (Sigma; 2 mg/ml in water) was homogenized by sonication for 6 minutes (Labsonic U, Braun). DNA-fragments were between 300 and 3,000 base pairs in size, as revealed by agarose gel analysis. Melting curve measurements were performed using the Cary 1E UV-visible spectrophotometer (Varian) as reported previously⁴⁾. Interaction with DNA is indicated by a change in melting temperature (ΔT_m [°C]).

Thin-layer Chromatography

TLC binding studies were performed with the 1D-TLC method on silica gel plates (Merck, HPTLC-ready-to-use-plates, silica gel RP-18 WF_{254S} on glass)⁴⁾. Pure substances (total amount per spot: 5 μ g) were analyzed for DNA-binding properties in a solvent system consisting of methanol-1 M aqueous ammonium acetate (4:1). Homogenized salmon sperm DNA (total amount per spot: 4 μ g) was spotted above the sample spots prior to chromatography.

Detection was performed by UV extinction or fluorescence (254 and 366 nm) as well as by colorization with staining reagents (e.g. anisaldehyde-H₂SO₄, orcinol reagent, Ehrlich's reagent, and blue tetrazolium reagent). Changes in R_f-values indicate an interaction between ligand and DNA and are expressed by the R_{f2}/R_{f1}-ratio in which R_{f1} represents the R_f-value without, and R_{f2} with DNA.

Investigated Low Molecular Weight Natural Substances

All compounds tested are natural products and derivatives, either commercially available, published or unpublished, which are part of a collection of natural products, located at the Hans-Knöll-Institute for Natural Products Research, Jena, Germany. The compounds were dissolved in a suitable organic solvent (e.g. methanol, acetone, chloroform or water) resulting in a 1 mg/ml concentrated stock solution which then was spotted onto the TLC-plates.

Results

Detection of DNA-binding Compounds

According to the 1D-TLC method described in the previous paper 470 selected natural products were examined. 5 μ g of each pure secondary metabolite dis-

solved in a suitable organic solvent was applied twice on a RP-18 WF_{254S}-TLC-plate: In the first lane without DNA and in a second lane together with a homogenized random-DNA sample (4 μ g, dissolved in water). After chromatography using the solvent system methanol-1 M aqueous ammonium acetate (4:1) the ratio of the R_f-values (R_{f2}/R_{f1}) of the natural product obtained with DNA (R_{f2}) and without DNA (R_{f1}) was determined. An interaction with DNA is indicated when the R_{f2}/R_{f1}-ratio decreases significantly below 1. Visualization of the metabolites was therefore realized by UV detection (254 and 366 nm) and/or staining with defined staining reagents. Because a chromatographic step on TLC is involved in the binding studies, it was possible to differentiate between DNA-binding properties of the natural product itself and impurities of the sample. The compounds examined were structurally classified according to the system introduced by BERDY⁵⁾ (see Table 1).

The results of the binding studies are summarized in Fig. 1 and Table 1. As expected among the quinones (hydroquinones, naphthoquinones, and anthraquinones; number of test compounds: n=15) a major percentage (66.6%) of compounds with DNA-binding properties was observed. Analogous results were obtained with the polyethers (71.4%; n=7), the macrolides (36.4%; n=11), and alkaloids with 19.3% (n=57). In contrast, with other *N*-heterocycles (3.1%; n=65), *O*-heterocycles (2.1%; n=47) and aromatic compounds (2.9%; n=34) only few compounds exhibited DNA-binding properties. None of the saccharides (mono-, di-, oligo- and polysaccharides; n=50) and amino acids (n=48) showed DNA affinity.

In order to give more detailed information, Table 2 provides an overview of the compounds found to interact with DNA, the obtained R_{f2}/R_{f1}-ratio, as well as known biological activities cited from literature. In order to prove the reliability of the method, about 5% of all compounds selected for our studies were known DNA binders. In all cases we established the DNA-binding properties independently by the 1D-TLC method, which *vice versa* proved the reliability of the biomolecular-chemical screening method. Other compounds identified as DNA-binders on TLC have been reported to exhibit antiviral (e.g. formycin B⁴²⁾ or puromycin⁴³⁾ or antitumor activity (e.g. echitamine¹¹⁾ or ivermectin²²⁾), or act as inhibitors of DNA synthesis (e.g. bacitracin²⁷⁾ or calcimycin³⁰⁾). The TLC-based binding towards DNA of these type of natural compounds implicates a possible interaction with DNA being responsible for the biological activity.

For a number of natural products examined here, DNA-binding was detected for the first time. The TLC based

Table 1. DNA-binding behaviour of selected natural products analysed by 1D TLC-binding studies.

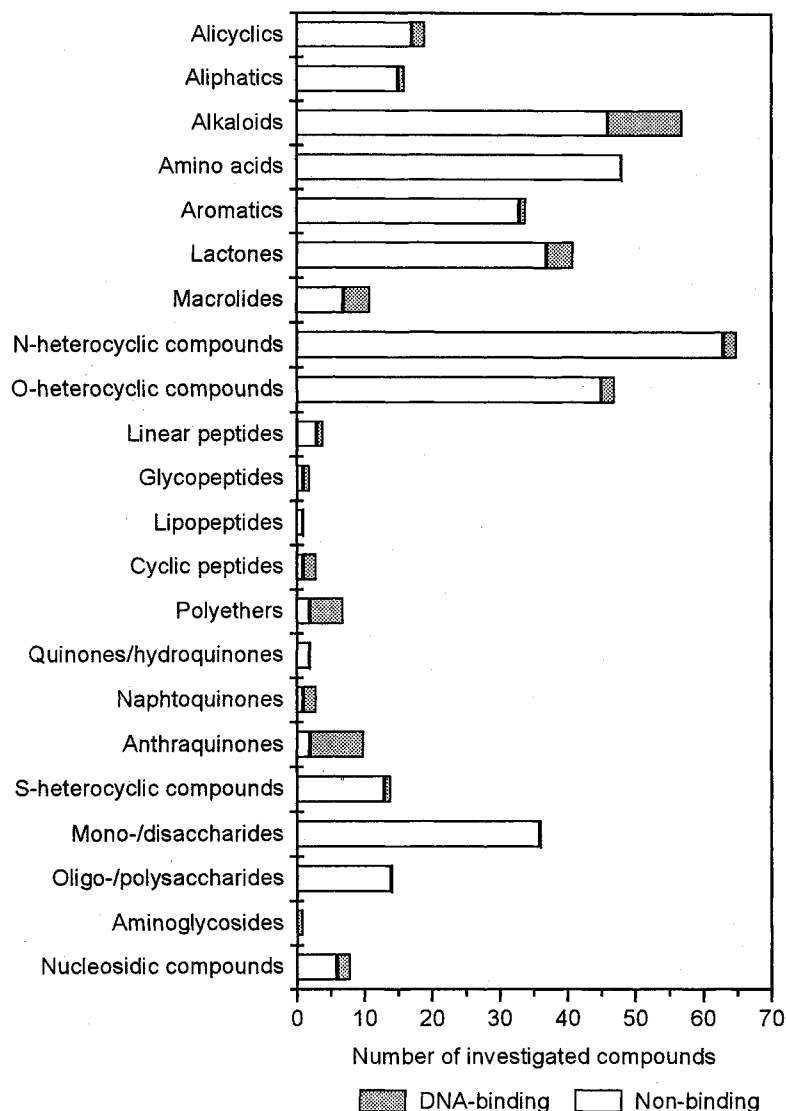
Chemical class	Number of tested compounds	DNA-binding
Alicyclics	19	2
Aliphatics	16	1
Alkaloids	57	11
Amino acids	48	0
Aromatics	34	1
Lactones	41	4
Macrolides	11	4
N-heterocyclic compounds	65	2
O-heterocyclic compounds	47	2
Peptides	10	4
Linear peptides	(4)	(1)
Glycopeptides	(2)	(1)
Lipopptides	(1)	(0)
Cyclic peptides	(3)	(2)
Polyether	7	5
Quinones	15	10
Quinones/hydroquinones	(2)	(0)
Naphthoquinones	(3)	(2)
Anthraquinones	(10)	(8)
S-heterocyclic compounds	14	1
Sugars	59	3
Mono- and disaccharides	(36)	(0)
Oligo- and polysaccharides	(14)	(0)
Aminoglycosides	(1)	(1)
Nucleosidic compounds	(8)	(2)
Total	470	50

findings were proven by comparative DNA melting curve measurements. For example, the cytotoxic effect of culmorin⁶⁾ produced by *Fusarium graminearum* might be explained by DNA-binding as discovered with the 1D-TLC method ($Rf_2/Rf_1=0.27$; $\Delta T_m=1.0^\circ\text{C}$). Surprisingly DNA-binding properties were found with compounds which have a defined biological activity seemingly unrelated to any DNA interaction. Examples are raubasine produced by *Rauwolfia* spp. ($Rf_2/Rf_1=0.87$; $\Delta T_m=1.0^\circ\text{C}$) which shows affinity to the α -2C adrenergic receptor¹³⁾, megalomicin from *Micromonospora megalomicea* ($Rf_2/Rf_1=0.66$; $\Delta T_m=1.0^\circ\text{C}$) an inhibitor of intra-golgi transport²³⁾, and decarestrictine D from *Penicillium corylophilum* and *P. simplicissimum* ($Rf_2/Rf_1=0.91$; $\Delta T_m=0.8^\circ\text{C}$), or deschlorothricin from *Streptomyces antibioticus* ($Rf_2/Rf_1=0.89$; $\Delta T_m=1.2^\circ\text{C}$), both inhibitors of *de novo* cholesterol biosynthesis^{17,21)}. Furthermore, affinity to DNA was discovered for a number of compounds for which no biological effect has been described up to now. Examples are 1-*O*-(3-hydroxybenzoyl)- β -D-4,5-dideoxy-4-aminoribopyranose ($Rf_2/Rf_1=0.84$; $\Delta T_m=0.8^\circ\text{C}$), 6-phenyl-3,5-diethyl-4-hydroxy-2*H*-pyran-2-on ($Rf_2/Rf_1=0.80$; $\Delta T_m=0.9^\circ\text{C}$) or yohimbin carboxylic acid ($Rf_2/Rf_1=0.70$; $\Delta T_m=1.1^\circ\text{C}$).

Discussion

By analysing a broad range of pure low molecular weight natural metabolites with the new TLC-based DNA binding

Fig. 1. Graphical overview on DNA-binding properties of selected natural products analyzed by 1D TLC-binding studies.



assay, the reliability of the screening approach has been proven. However, in a few cases the 1D-TLC method may fail to detect a defined DNA affinity for some compounds. If compounds exhibit a very low R_f -value ($R_f < 0.1$) in the standard solvent system, it will be impossible to detect changes in the R_f -values. For these type of compounds a suitable solvent system should be developed. But within different solvent systems the values of the R_{f2}/R_{f1} -ratio are not comparable. Some compounds may be not detectable by the routinely applied UV-extinction/fluorescence (256 and 366 nm) or anisaldehyde- H_2SO_4 . Here, more special staining reagents, *e.g.* orcinol reagent, Ehrlich's reagent or blue tetrazolium reagent, or densitometric read-outs have to

be used to open the window to structural diversity.

DNA-binding properties of already known DNA-binders such as intercalators and groove-binders belonging to different chemical classes could be independently identified by the 1D-TLC method. Additional prove for the applicability of the method derived from DNA-melting curve analysis and CD-measurements of DNA applied under solvent conditions as on the TLC⁽⁴⁾. Thus, TLC-based DNA-binding detection constitutes a reliable and easy-to-handle method which can advantageously be applied in preliminary binding studies and, for pure compounds, allows a high throughput analysis due to parallelization of TLC. Future application will be directed towards the

Table 2. Survey of detected DNA-binding natural products, Rf_2/Rf_1 -ratios as derived from the 1D-TLC method and their reported biological activities.

Chemical class Compound	Rf_2/Rf_1^a	Biological activity
Alicyclics		
Culmorin	0.27	Cytotoxic ⁶⁾
Gabosin F	0.82	— ^b
Aliphatics		
Cyclopentan-1-on-3-methylen-1'-methylene-2-carboxylic acid	0.87	— ^b
Alkaloids		
Cephalotaxine	0.91	DNA-binding ⁷⁾
Chelerythrine	0	Inhibition of Protein Kinase C, DNA-binding ^{8,9)}
Conodurine	0.80	Antimicrobial ¹⁰⁾
Echitamine	0.85	Antitumor ¹¹⁾
Pheanthine	0.80	— ^b
Quebrachamine	0.80	α -2 Receptor antagonist ¹²⁾
Raubasine	0.87	α -2C Receptor antagonist ¹³⁾
Sanguinarine	0.80	DNA-binding ¹⁴⁾
Tetrandrine	0.87	Anti-inflammatory, antineoplastic ¹⁵⁾
Tomatidine	0.50	Cytotoxic ¹⁶⁾
Yohimbine carboxylic acid	0.70	— ^b
Aromatics		
2-Carbinol-3,6-methyl-5-methoxy-anisol	0.87	— ^b
Lactones		
Decarestrictine D	0.91	Inhibition of cholesterol biosynthesis ¹⁷⁾
5-Hexyldihydro-2(3H)-furanone	0.88	Antifungal ¹⁸⁾
Inamycin	0.88	DNA-binding ¹⁹⁾
Trioxsalen	0.70	DNA-binding ²⁰⁾
Macrolides		
Deschlorothricin	0.89	Inhibition of cholesterol biosynthesis ²¹⁾
Ivermectin	0.90	Anthelmintic, antitumor ²²⁾
Megalomicin	0.66	Inhibition of intra-golgi-transport ²³⁾
Rutamycin A	0.91	ATPase inhibition ²⁴⁾
N-heterocyclic compounds		
3-Ethyl-4-hydroxy-6,8-dimethyl-1,8-dihydro-1,2,6,8,9-pentaazacyclopentanaphtalen-5,7-dion	0.89	— ^b
8-Methyl-5-phenyl-3-[(8-phenyl-1,2,3,4-tetrahydro-isoquinolen-5-yl)-hydrazone]-1,3,6,7,8,9-hexahydropyrrolo-[3,2]-isoquinolin-2-on	0.81	— ^b

Table 2. (Continued)

Chemical class Compound	Rf ₂ /Rf ₁ ^a	Biological activity
<i>O</i> -heterocyclic compounds		
Cycloserine	0.91	DNA-binding ²⁵⁾
6-Phenyl-3,5-diethyl-4-hydroxy-2 <i>H</i> -pyran-2-on	0.80	— ^b
Peptides		
Actinomycin D	0	DNA-binding ²⁶⁾
Bacitracin	0.80	Inhibition of DNA synthesis ²⁷⁾
Distamycin	0.80	DNA-binding ²⁸⁾
Phleomycin	0.70	DNA-binding ²⁹⁾
Polyether		
Calcimycin	0.88	Inhibition of DNA synthesis ³⁰⁾
Monensin A	0.50	Antiviral ³¹⁾
Nigericin	0.88	Antiviral ³²⁾
Salinomycin	0.92	Antiviral ³¹⁾
X-206	0.75	Antiviral ³³⁾
Quinones		
Aclacinomycin	0.66	DNA-binding ³⁴⁾
Aklanonic acid	0.82	DNA-binding ³⁵⁾
Aklavinone	0.77	DNA-binding ³⁶⁾
2-Chloro-3(2-ethoxycarbonyl-3-benzylamino-β-phenylethylen)-[1,4]-naphtoquinone	0.88	— ^b
Daunomycin	0	DNA-binding ³⁷⁾
Doxorubicin	0	DNA-binding ³⁷⁾
Iremycin	0.73	DNA-binding ³⁸⁾
Nogalamycin	0	DNA-binding ³⁹⁾
Steffimycin	0.88	DNA-binding ⁴⁰⁾
Vitamin K	0.92	DNA-binding ⁴¹⁾
<i>S</i> -heterocyclic compounds		
2-Methyl-3-ethoxycarbonyl-4-oxo-5-dimethylaminomethylen-thiophen	0.70	— ^b
Sugars		
Formycin B	0.90	Antiviral ⁴²⁾
1- <i>O</i> -(3-Hydroxybenzoyl)-β-D-4,5-dideoxy-4-aminoribopyranose	0.84	— ^b
Puromycin	0.84	Antiviral, antitumor ⁴³⁾

^a Changes in Rf-values are expressed by the Rf₂/Rf₁-ratio in which Rf₁ represents the Rf-value without, and Rf₂ represents the Rf-value with DNA;

^b No biological activity reported up to now.

examination of libraries from combinatorial synthesis approaches⁴⁴).

Applying the 1D-TLC binding studies it was possible to detect DNA-binding properties of a number of compounds that were not known before. As the identified compounds belong to different chemical classes, the results indicate the potential of the presented method not only to detect strong DNA-binding molecules of limited structural variety.

Acknowledgements

We would like to thank U. VALENTIN for excellent technical assistance. This research was supported by a DECHEMA-scholarship (grant to C.M.).

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